

Sulfur-Containing Spiroketal Glycosides from *Breynia fruticosa*

Dahai Meng, Wenliang Chen, and Weimin Zhao*

Shanghai Institute of Materia Medica, Shanghai Institutes for Biological Sciences, Chinese Academy of Sciences, Shanghai 201203, People's Republic of China

Received December 19, 2006

Phytochemical investigation of the chemical constituents of the aerial parts of *Breynia fruticosa* afforded 10 sulfur-containing spiroketal glycosides, including eight new compounds (**1–8**) and two known compounds, epibreynin B (**9**) and breynin B (**10**). Epibreynin B (**9**) was previously reported as a semisynthetic oxidation product of breynin A and is now reported from a natural source for the first time. The structures of **1–10** were determined on the basis of spectroscopic and chemical methods.

The genus *Breynia* (Euphorbiaceae) is comprised of 25 species of plants, seven of which are found in China. *Breynia fruticosa* (L.) Hook. f. is the one distributed abundantly in the south of China and has been used as a Chinese folk medicine for the treatment of chronic bronchitis and wounds. Previous phytochemical studies on *Breynia* plants led to the identification of sulfur-containing spiroketal glycosides,^{1–4} alkaloids,^{5,6} and terpenoid and phenolic glycosides,⁷ along with several other components.^{8,9} Among those identified components, breynins A and B, the two sulfur-containing spiroketal glycosides, exhibited significant oral hypocholesterolemic activity.^{1–4} As a part of our ongoing program on the discovery of new bioactive components from terrestrial plants, the aerial parts of *B. fruticosa* were studied systematically. This paper describes the isolation and structural elucidation of eight new sulfur-containing spiroketal glycosides (**1–8**), along with two known compounds, epibreynin B (**9**) and breynin B (**10**). Epibreynin B (**9**) was previously reported as a semisynthetic oxidation product of breynin A⁴ and is now reported from a natural source. The structural identification of these compounds is based on the known skeleton, breynolide, whose structure was secured via single-crystal X-ray analysis^{4,10} and also on the basis of spectroscopic and chemical methods.

Results and Discussion

Powdered, air-dried aerial parts of *Breynia fruticosa* (30.0 kg) were percolated with 95% EtOH ($\times 3$) at room temperature. The filtrate was concentrated to dryness *in vacuo* and then suspended in 20% EtOH. After filtration of the precipitated chlorophyll and evaporation of EtOH from the filtrate, the aqueous residue was extracted with CHCl₃ and *n*-BuOH, successively. Ten sulfur-containing compounds were afforded from the *n*-BuOH fraction through a series of column chromatographies including silica gel and Sephadex LH-20 columns, HPLC, and PTLC steps.

Compound **1** showed a pseudomolecular ion at *m/z* 921.3094 [M + H]⁺ in the positive-ion mode HRESIMS and at *m/z* 943.6 [M + Na]⁺ in the positive ESIMS. When considered in conjunction with its ¹³C NMR data, it indicated a molecular formula of C₄₀H₅₆O₂₂S. The assignment of ¹H and ¹³C NMR data (Tables 1 and 2) was based on HSQC, HMBC, ¹H–¹H COSY (Figure 1), and TOCSY spectra. In the ¹³C NMR spectrum of **1**, the presence of three sugar units was indicated, along with 22 carbons for the aglycone, breynogenin (**11**), which was identified by comparison of spectroscopic data with literature values.^{4,11} The identities of the sugar moieties were elucidated from extensive analysis of the NMR data (¹³C, ¹H, ¹H–¹H COSY, TOCSY, HMBC, ROESY, and HSQC) of the carbohydrate chain. From the three anomeric carbons

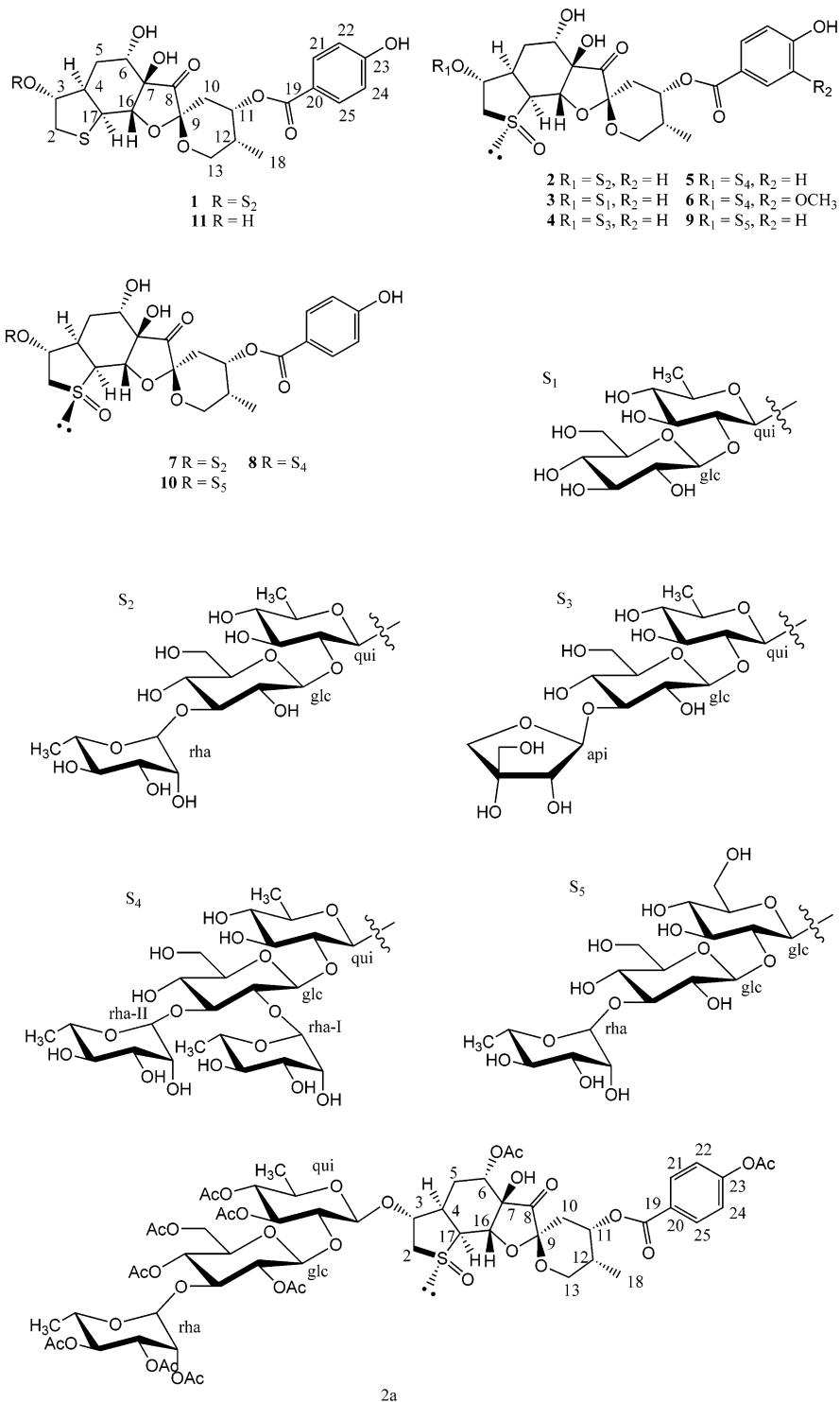
(δ_{C} 106.9, 104.1, and 102.9) and three anomeric protons [δ_{H} 4.08 (1H, d, $J = 7.4$ Hz), 4.42 (1H, d, $J = 7.7$ Hz), and 5.10 (1H, d, $J = 1.5$ Hz)] and other characteristic NMR resonances, the sugar units were identified as β -quinovopyranose, β -glucopyranose, and α -rhamnopyranose, which was further confirmed by strong NOEs between H-1_{qui} and H-3_{qui}, H-5_{qui}; H-1_{glc} and H-3_{glc}, H-5_{glc}; and H-1_{rha} and H-2_{rha} in its ROESY spectrum.¹² The ¹³C NMR chemical shift of C-3 (δ_{C} 91.8) suggested that **1** had a glycosyl linkage at C-3. The sugar sequence and their linkage sites were derived from the HMBC experiment, correlating H-1_{qui} (δ_{H} 4.42)/C-3 (δ_{C} 91.8), H-1_{glc} (δ_{H} 4.08)/C-2_{qui} (δ_{C} 86.2), and H-1_{rha} (δ_{H} 5.10)/C-3_{glc} (84.6). The remaining HMBC correlations are shown in Figure 1. Other COSY and HMBC correlations (Figure 1) were consistent with the aglycone of **1** being breynogenin (**11**). Thus, compound **1** was determined as breynogenin 3-*O*- α -rhamnopyranosyl-(1 \rightarrow 3)- β -glucopyranosyl-(1 \rightarrow 2)- β -quinovopyranoside, trivial name breynin C.

Compound **2** showed a quasimolecular ion [M + Na]⁺ at *m/z* 959.2845 in the positive HRESIMS, indicative of a molecular formula of C₄₀H₅₆O₂₃S. When its ¹H and ¹³C NMR data (Tables 1 and 2) were compared with those of **1**, the deshielding of C-2 (from δ_{C} 36.3 to 62.1) and C-17 (from δ_{C} 47.3 to 62.8) of the aglycone was observed. Comparison of their NMR data with those in the literature showed that the aglycone of **2** was β -sulfoxidebreynogenin.^{4,13} In order to further confirm its structure, acetylation of **2** was performed to give the undecaacetate **2a**. Its ¹H NMR spectrum displayed more clearly separated sugar proton resonances. Analysis of its ¹H–¹H COSY, TOCSY, and ROESY spectra enabled assignment of all sugar protons of **2a**. The chemical shifts of the shielded H-2_{qui} (δ_{H} 3.58) and H-3_{glc} (δ_{H} 3.65) in the ¹H NMR spectrum of **2a**, and also the NOE correlations between H-1_{qui} (δ_{H} 4.36) and H-3 (δ_{H} 3.89), between H-1_{glc} (δ_{H} 4.49) and H-2_{qui} (δ_{H} 3.58), and between H-1_{rha} (δ_{H} 4.72) and H-3_{glc} (δ_{H} 3.65), indicated identical glycosyl moieties in **1** and **2**. Analysis of the ¹H–¹H COSY, TOCSY, and ROESY spectra of **2a** also confirmed the structure of its aglycone. Accordingly, compound **2** was characterized as β -sulfoxidebreynogenin 3-*O*- α -rhamnopyranosyl-(1 \rightarrow 3)- β -glucopyranosyl-(1 \rightarrow 2)- β -quinovopyranoside (epibreynin D).

Compounds **3**, **4**, and **5** were assigned the elemental compositions C₃₄H₄₆O₁₉S, C₃₉H₅₄O₂₃S, and C₄₆H₆₆O₂₇S, respectively, from analysis of HRESIMS and NMR data. On the basis of HSQC, TOCSY, ¹H–¹H COSY, and HMBC spectra, all ¹H and ¹³C resonances of **3**, **4**, and **5** were assigned as shown in Tables 1 and 2. Comparison of the NMR data of **3**, **4**, and **5** with **2** showed that these compounds possessed the same aglycone but differed glycosyl moieties. The anomeric resonances at δ_{H} 4.12 (1H, d, $J = 7.2$ Hz) and 4.44 (1H, d, $J = 7.8$ Hz) in the ¹H NMR spectrum suggested that **3** had two sugar units, which were determined to be β -glucopyranose and β -quinovopyranose by a combination of ¹H–¹H COSY, HSQC, and ROESY experiments and a detailed comparison

* To whom correspondence should be addressed. Tel & Fax: +86 21 50806052. E-mail: wimzhao@mail.shnc.ac.cn.

Chart 1



of NMR data with those of **2**. The HMBC spectrum showed cross-peaks between the anomeric protons H-1_{qui} with C-3, C-2_{qui}, C-3_{qui}, and H-1_{glc} with C-2_{qui}, C-2_{glc}, C-3_{glc}, and C-5_{glc}, respectively. These interactions supported the presence of a sugar moiety, glc-(1→2)-qui, and its linkage to C-3 of β -sulfoxidebreynogenin. From the above evidence, the structure of **3** was established as β -sulfoxidebreynogenin 3-O- β -glucopyranosyl-(1→2)- β -quino-vopyranoside.

The NMR data of **4** were similar to those of **3**, except for resonances of one additional sugar unit. On the basis of ¹H-¹H COSY, HSQC, and TOCSY spectra, all sugar protons and carbons could be clearly assigned (Tables 1 and 2), from which the extra

sugar unit was concluded to be β -apiofuranose.^{12,14} The interglycosidic linkages were determined from the following HMBC correlations: H-1_{qui} (δ_H 4.42)/C-3 (δ_C 88.2); H-1_{glc} (δ_H 4.12)/C-2_{qui} (δ_C 86.1); and H-1_{api} (δ_H 5.22)/C-3_{glc} (δ_C 86.4). On the basis of the above evidence, compound **4** was identified as β -sulfoxidebreynogenin 3-O- β -apiofuranosyl-(1→3)- β -glucopyranosyl-(1→2)- β -quinovopyranoside.

The NMR data of compound **5** were similar to those of **2** (Tables 1 and 2), except for the presence of one additional sugar unit, which was concluded to be α -rhamnopyranose from the characteristic NMR spin patterns. The monosaccharide sequence was determined by analysis of HMBC correlations. Cross-peaks were observed

Table 1. ^{13}C NMR Data of **1–8** (100 MHz, CD_3OD)

position	1	2	3	4	5	6	7	8
2	36.3 (t)	62.1 (t)	62.1 (t)	61.9 (t)	61.7 (t)	61.7 (t)	57.7 (t)	57.9 (t)
3	91.8 (d)	88.2 (d)	88.1 (d)	88.2 (d)	86.2 (d)	86.7 (d)	87.9 (d)	86.1 (d)
4	39.7 (d)	40.3 (d)	40.2 (d)	40.2 (d)	40.1 (d)	40.1 (d)	38.8 (d)	38.8 (d)
5	27.7 (t)	29.0 (t)	29.0 (t)	29.0 (t)	29.0 (t)	28.9 (t)	29.6 (t)	29.8 (t)
6	71.2 (d)	71.8 (d)	71.9 (d)	71.9 (d)	71.0 (d)	71.2 (d)	70.6 (d)	70.5 (d)
7	76.0 (s)	73.8 (s)	73.7 (s)	73.7 (s)	73.8 (s)	73.7 (s)	75.5 (s)	75.6 (s)
8	213.7 (s)	210.3 (s)	210.9 (s)	210.5 (s)	210.4 (s)	210.3 (s)	212.7 (s)	212.6 (s)
9	100.1 (s)	101.4 (s)	101.5 (s)	101.5 (s)	101.2 (s)	101.3 (s)	100.9 (s)	100.7 (s)
10	32.9 (t)	33.0 (t)	32.9 (t)	33.0 (t)	32.3 (t)	32.8 (t)	32.7 (t)	32.2 (t)
11	70.2 (d)	70.3 (d)	70.2 (d)					
12	34.7 (d)	34.6 (d)	34.6 (d)	34.6 (d)	34.4 (d)	34.5 (d)	34.6 (d)	34.4 (d)
13	64.1 (t)	64.3 (t)	64.0 (t)	64.4 (t)	64.6 (t)	64.4 (t)	64.7 (t)	65.2 (t)
16	77.4 (d)	74.7 (d)	74.4 (d)	74.7 (d)	74.7 (d)	74.7 (d)	75.4 (d)	75.6 (d)
17	47.3 (d)	62.8 (d)	62.7 (d)	62.7 (d)	62.2 (d)	62.5 (d)	71.0 (d)	70.0 (d)
18	13.1 (q)	13.2 (q)	13.2 (q)	13.2 (q)	13.0 (q)	13.0 (q)	13.0 (q)	12.7 (q)
19	168.0 (s)							
20	123.0 (s)	123.2 (s)	122.9 (s)	122.8 (s)	123.0 (s)	116.9 (s)	123.0 (s)	122.9 (s)
21	133.7 (d)	133.6 (d)	133.6 (d)	133.6 (d)	133.5 (d)	126.2 (d)	133.6 (d)	133.5 (d)
22	117.0 (d)	116.9 (d)	116.9 (d)	117.0 (d)	116.7 (d)	116.8 (d)	117.0 (d)	116.8 (d)
23	163.8 (s)	163.6 (s)	164.1 (s)	164.2 (s)	163.6 (s)	149.0 (s)	163.8 (s)	163.8 (s)
24	117.0 (d)	116.9 (d)	116.9 (d)	117.0 (d)	116.7 (d)	149.0 (s)	117.0 (d)	116.8 (d)
25	133.7 (d)	133.6 (d)	133.6 (d)	133.6 (d)	133.5 (d)	113.8 (d)	133.7 (d)	133.5 (d)
24-OCH ₃	qui							
1	104.1 (d)	103.4 (d)	103.3 (d)	103.4 (d)	102.2 (d)	102.5 (d)	103.9 (d)	102.7 (d)
2	86.2 (d)	85.8 (d)	85.8 (d)	86.1 (d)	82.2 (d)	83.3 (d)	85.0 (d)	81.2 (d)
3	77.9 (d)	77.7 (d)	77.9 (d)	77.8 (d)	78.1 (d)	78.3 (d)	78.0 (d)	78.4 (d)
4	76.5 (d)	76.4 (d)	76.3 (d)	76.4 (d)	76.7 (d)	76.7 (d)	76.5 (d)	77.0 (d)
5	73.7 (d)	73.6 (d)	74.0 (d)	73.8 (d)	73.6 (d)	73.6 (d)	73.7 (d)	73.7 (d)
6	18.4 (q)	18.4 (q)	18.4 (q)	18.4 (q)	18.1 (q)	18.1 (q)	18.3 (q)	18.1 (q)
	glc							
1	106.9 (d)	106.9 (d)	107.0 (d)	106.9 (d)	104.0 (d)	104.5 (d)	106.5 (d)	103.7 (d)
2	77.0 (d)	77.0 (d)	76.7 (d)	76.0 (d)	79.8 (d)	79.5 (d)	77.0 (d)	79.8 (d)
3	84.6 (d)	84.3 (d)	78.0 (d)	86.4 (d)	87.6 (d)	88.0 (d)	84.2 (d)	87.6 (d)
4	69.4 (d)	69.4 (d)	70.9 (d)	69.3 (d)	70.1 (d)	69.9 (d)	69.7 (d)	70.5 (d)
5	77.5 (d)	77.7 (d)	77.7 (d)	77.4 (d)	77.3 (d)	77.1 (d)	77.8 (d)	77.7 (d)
6	61.9 (t)	62.0 (t)	62.0 (t)	61.9 (t)	62.1 (t)	61.8 (t)	62.2 (t)	62.5 (t)
	rha	rha	api	rha-I	rha-I	rha	rha	rha-I
1	102.9 (d)	102.8 (d)		111.7 (d)	103.0 (d)	103.0 (d)	102.8 (d)	102.9 (d)
2	72.5 (d)	72.5 (d)		78.2 (d)	72.1 (d)	72.2 (d)	72.5 (d)	72.2 (d)
3	72.6 (d)	72.6 (d)		80.7 (s)	72.3 (d)	72.3 (d)	72.6 (d)	72.4 (d)
4	74.3 (d)	74.3 (d)		75.3 (t)	73.6 (d)	74.1 (d)	74.3 (d)	74.0 (d)
5	70.5 (d)	70.4 (d)		65.3 (t)	70.8 (d)	70.8 (d)	70.4 (d)	70.6 (d)
6	18.2 (q)	18.2 (q)			18.4 (q)	18.4 (q)	18.2 (q)	18.5 (q)
1					103.9 (d)	104.0 (d)		104.0 (d)
2					72.6 (d)	72.6 (d)		72.7 (d)
3					72.4 (d)	72.4 (d)		72.5 (d)
4					73.8 (d)	73.8 (d)		74.3 (d)
5					71.7 (d)	71.8 (d)		71.0 (d)
6					18.4 (q)	18.5 (q)		18.4 (q)

between H-1_{qui} (δ_{H} 4.42)/C-3 (δ_{C} 86.2); H-1_{glc} (δ_{H} 4.58)/C-2_{qui} (δ_{C} 82.2); H-1_{rha-I} (δ_{H} 5.05)/C-2_{glc} (δ_{C} 79.8); and H-1_{rha-II} (δ_{H} 4.93)/C-3_{glc} (δ_{C} 87.6). Therefore, the structure of compound **5** was established as β -sulfoxidebreynogenin 3-O-[α -rhamnopyranosyl-(1 \rightarrow 3)]- α -rhamnopyranosyl-(1 \rightarrow 2)- β -glucopyranosyl-(1 \rightarrow 2)- β -quino-vopyranoside. Compounds **3–5** are new and have been assigned the trivial names epibreynins E (**3**), F (**4**), and G (**5**), respectively.

The HRESIMS of compound **6** showed a sodiated molecular ion at m/z 1135.3535 [M + Na]⁺, which, in conjunction with ^{13}C NMR data, established the molecular formula of $\text{C}_{47}\text{H}_{68}\text{O}_{28}\text{S}$. On the basis of its HSQC, ^1H - ^1H COSY, and TOCSY NMR spectra, all resonances of compound **6** were assigned as shown in Tables 1 and 2. Comparison of the NMR data for compounds **5** and **6** suggested the presence of the same sugar moieties. The ^{13}C NMR spectrum showed 47 resonances, of which 23 were attributed to the aglycone moiety, including a methoxy group at δ_{C} 57.0. In the ^1H NMR spectrum, resonances confirming the substitution pattern of the benzoyl moiety [δ_{H} 7.72 (1H, dd, J = 8.3, 1.5 Hz); δ_{H} 7.56 (1H, d, J = 1.5 Hz); δ_{H} 6.88 (1H, d, J = 8.3 Hz)] were observed. The correlation of a methoxy resonance at δ_{H} 3.89 with a carbon

at δ_{C} 149.0 in the HMBC spectrum indicated a methoxy group at C-24, which was further confirmed by correlations between H-21 (δ_{H} 7.72) and C-19 (δ_{C} 168.0), C-23 (δ_{C} 149.0), C-25 (δ_{C} 113.8); H-22 (δ_{H} 6.88) and C-20 (δ_{C} 116.9), C-24 (δ_{C} 149.0); and H-25 (δ_{H} 7.56) and C-19 (δ_{C} 168.0), C-21 (δ_{C} 126.2), C-23 (δ_{C} 149.0) in the HMBC spectrum (Figure 1). The assignments of the sugar moieties and linkages were also confirmed from the HMBC and COSY spectra. Accordingly, compound **6** was determined as 24-methoxy- β -sulfoxidebreynogenin 3-O-[α -rhamnopyranosyl-(1 \rightarrow 3)]- α -rhamnopyranosyl-(1 \rightarrow 2)- β -glucopyranosyl-(1 \rightarrow 2)- β -quino-vopyranoside and named epibreynin H.

Compound **7** exhibited the same elemental formula ($\text{C}_{40}\text{H}_{56}\text{O}_{23}\text{S}$) as **2**. The ^1H and ^{13}C NMR spectra of **7** were similar to those of **2**, except for the significant differences of the chemical shifts of C-2 [δ_{C} 57.7 (**7**), 62.1 (**2**)], C-17 [δ_{C} 71.0 (**7**), 62.8 (**2**)], and C-7 [δ_{C} 75.5 (**7**), 73.8 (**2**)]. Detailed comparison of ^{13}C NMR data of **7** and **2** with those reported revealed the aglycones of **2** and **7** to be isomers at the sulfoxide position.⁴ That is, compound **7** possessed α -sulfoxidebreynogenin as its aglycone. Analysis of NMR data (^{13}C , ^1H , ^1H - ^1H COSY, HSQC, TOCSY, HMBC, and ROESY) showed

Table 2. ^1H NMR Data of **1–8** (400 MHz, CD_3OD)

position	1	2	3	4	5	6	7	8
2 α	3.20 (m)	4.20 (d, 14.9)	4.26 (d, 14.9)	4.26 (d, 15.0)	4.16 (d, 15.0)	4.26 (d, 14.7)	3.78 (dd, 15.4, 5.7)	3.85 (dd, 14.0, 5.7)
2 β	3.04 (dd, 11.7, 3.3)	3.12 (m)	3.14 (m)	3.13 (m)	3.15 (m)	3.12 (m)	3.25 (m)	3.10 (m)
3	4.32 (br d, 3.3)	4.67 (br d, 4.2)	4.71 (br d, 4.8)	4.60 (br d, 3.7)	4.72 (br s)	4.71 (br s)	4.44 (br d, 4.9)	4.50 (br s)
4	2.85 (dt, 13.3, 4.3)	3.16 (m)	3.18 (m)	3.21 (m)	3.17 (m)	3.20 (m)	3.14 (m)	3.20 (m)
5 α	1.65 (dt, 13.7, 3.7)	1.81 (m)	1.64 (m)	1.80 (m)	1.82 (m)	1.84 (m)	1.25 (m)	1.24 (br d, 14.2)
5 β	1.50 (dt, 2.3, 13.7)	1.81 (m)	1.56 (m)	1.76 (m)	1.82 (m)	1.80 (m)	1.70 (dt, 14.2, 3.6)	1.70 (dt, 14.2, 3.5)
6	3.92 (m)	3.92 (m)	3.84 (m)	3.83 (m)	3.92 (m)	3.90 (m)	3.98 (m)	3.93 (m)
10	2.05 (br d, 3.8)	2.10 (m)	2.04 (m)	2.15 (m)	2.06 (br d, 3.5)	2.14 (m)	2.12 (br d, 4.1)	2.15 (m)
11	5.48 (m)	5.38 (m)	5.40 (m)	5.40 (m)	5.42 (m)	5.46 (m)	5.38 (m)	5.50 (m)
12	2.15 (m)	2.18 (m)	2.18 (m)	2.20 (m)	2.22 (m)	2.23 (m)	2.17 (m)	2.18 (m)
13 α	3.58 (m)	4.05 (m)	4.06 (t, 11.0)	4.02 (t, 11.2)	4.04 (m)	4.06 (m)	3.98 (m)	3.90 (m)
13 β	3.98 (m)	3.65 (m)	3.65 (dd, 11.0, 4.5)	3.65 (dd, 11.2, 4.4)	3.73 (m)	3.69 (m)	3.68 (m)	3.80 (m)
16	4.37 (br s)	4.90 (br s)	4.86 (br s)	4.89 (br s)	4.93 (br s)	4.91 (br s)	4.87 (br s)	4.75 (br s)
17	4.19 (br d, 5.0)	3.94 (m)	3.90 (m)	3.92 (m)	3.98 (m)	3.96 (m)	3.95 (m)	4.08 (m)
18	0.88 (d, 6.9)	0.88 (d, 6.9)	0.89 (d, 6.8)	0.90 (d, 6.8)	0.92 (d, 6.7)	0.93 (d, 7.0)	0.96 (d, 6.9)	0.98 (d, 6.9)
21	8.05 (d, 8.8)	8.05 (d, 8.4)	8.05 (d, 8.6)	8.03 (d, 8.7)	8.00 (d, 8.3)	7.72 (dd, 8.3, 1.5)	8.05 (d, 8.7)	8.00 (d, 8.7)
22	6.90 (d, 8.8)	6.88 (d, 8.4)	6.90 (d, 8.6)	6.88 (d, 8.7)	6.86 (d, 8.3)	6.88 (d, 8.3)	6.90 (d, 8.7)	6.90 (d, 8.7)
24	6.90 (d, 8.8)	6.88 (d, 8.4)	6.90 (d, 8.6)	6.88 (8.7)	6.86 (d, 8.3)	6.90 (d, 8.7)	6.90 (d, 8.7)	6.90 (d, 8.7)
25	8.05 (d, 8.8)	8.05 (d, 8.4)	8.05 (d, 8.6)	8.03 (8.7)	8.00 (d, 8.3)	7.56 (d, 1.5)	8.05 (d, 8.7)	8.00 (d, 8.7)
24-OCH ₃	qui	qui	qui	qui	qui	qui	qui	qui
1	4.42 (d, 7.7)	4.42 (d, 7.7)	4.44 (d, 7.8)	4.42 (d, 7.7)	4.42 (d, 7.1)	4.42 (d, 7.5)	4.46 (d, 7.5)	4.46 (d, 7.5)
2	3.36 (m)	3.28 (m)	3.30 (m)	3.30 (m)	3.47 (m)	3.41 (m)	3.45 (m)	3.58 (m)
3	3.45 (m)	3.45 (t, 9.2)	3.46 (m)	3.47 (m)	3.50 (m)	3.48 (m)	3.48 (m)	3.52 (m)
4	3.14 (t, 8.2)	3.14 (m)	3.15 (m)	3.13 (m)	3.06 (m)	3.1 (m)	3.14 (m)	3.04 (m)
5	3.31 (m)	3.33 (m)	3.34 (m)	3.38 (m)	3.31 (m)	3.33 (m)	3.32 (m)	3.25 (m)
6	1.35 (d, 6.1)	1.30 (d, 6.1)	1.31 (d, 6.1)	1.25 (d, 6.4)	1.24 (d, 6.0)	1.29 (d, 6.5)	1.28 (d, 6.0)	1.24 (d, 6.6)
glc	glc	glc	glc	glc	glc	glc	glc	glc
1	4.08 (d, 7.4)	4.05 (d, 7.6)	4.12 (d, 7.2)	4.12 (d, 7.8)	4.58 (d, 7.6)	4.48 (d, 6.9)	4.18 (d, 7.7)	4.70 (d, 7.2)
2	3.23 (m)	3.24 (m)	3.16 (m)	3.30 (m)	3.38 (m)	3.39 (m)	3.23 (m)	3.48 (m)
3	3.25 (m)	3.31 (m)	3.22 (m)	3.32 (m)	3.44 (m)	3.40 (m)	3.32 (m)	3.48 (m)
4	3.30 (m)	3.30 (m)	3.28 (m)	3.33 (m)	3.44 (m)	3.41 (m)	3.34 (m)	3.42 (m)
5	2.38 (dt, 9.2, 2.6)	2.48 (m)	2.46 (dt, 9.2, 2.6)	2.43 (dt, 9.6, 2.4)	2.76 (br d, 9.6, 9.6)	2.52 (br s)	2.62 (m)	2.95 (m)
6	3.46 (m); 3.62 (m)	3.52 (m); 3.57 (m)	3.50 (m); 3.57 (m)	3.52 (m); 3.56 (m)	3.52 (2H, m)	3.46 (2H, m)	3.56 (m); 3.66 (m)	3.58 (m); 3.66 (m)
rha	rha	api	rha-I	rha-I	rha-I	rha	rha-I	rha-I
1	5.10 (d, 1.5)	5.15 (d, 1.5)		5.22 (d, 3.4)	5.05 (br s)	5.05 (br s)	5.15 (d, 1.3)	5.05 (br s)
2	3.92 (m)	3.93 (m)		4.00 (m)	3.88 (m)	3.85 (m)	3.96 (m)	3.86 (m)
3	3.68 (dd, 9.5, 3.3)	3.67 (m)			3.70 (m)	3.71 (m)	3.72 (m)	3.72 (m)
4	3.40 (t, 9.6)	3.38 (t, 9.5)		3.82 (d, 10.0, 4.11 (m))	3.42 (m)	3.42 (m)	3.40 (m)	3.40 (m)
5	3.94 (m)	3.98 (m)		3.60 (m)	4.03 (m)	4.04 (m)	3.94 (m)	4.05 (m)
6	1.25 (d, 6.3)	1.27 (d, 6.2)			1.27 (d, 6.2)	1.31 (d, 6.1)	1.26 (d, 6.0)	1.30 (d, 6.4)
rha-II		rha-II	rha-II		4.93 (br s)	4.88 (br s)		rha-II
1					3.90 (m)	3.90 (m)		4.90 (br s)
2					3.67 (m)	3.65 (m)		3.90 (m)
3					3.43 (m)	3.42 (m)		3.65 (m)
4					3.96 (m)	3.95 (m)		3.42 (m)
5					1.25 (d, 6.0)	1.25 (d, 6.2)		3.98 (m)
6								1.26 (d, 6.2)

that compound **7** had the same saccharide chain as those of **1** and **2**. The correlations of an anomeric proton at H-1_{qui} (δ_H 4.46, $^3J = 7.5$ Hz) with C-3 (δ_C 87.9) and that of H-3 (δ_H 4.44) with C-1_{qui} (δ_C 103.9) located the sugar moiety at C-3. The structure of **7** was defined as α -sulfoxidebreyningenin 3-*O*- α -rhamnopyranosyl-(1 \rightarrow 3)- β -glucopyranosyl-(1 \rightarrow 2)- β -quinovopyranoside (breyinin D).

The molecular formula of compound **8** was determined as $C_{46}H_{66}O_{27}S$ from a pseudomolecular ion at m/z 1105.3359 [$M + \text{Na}^+$] in the positive-ion mode HRESIMS and at m/z 1081.3 [$M - \text{H}^-$] in negative-ion mode ESIMS. It exhibited the same elemental formula as **5**. On the basis of HSQC, TOCSY, ^1H - ^1H COSY, and

HMBC spectra, all ^1H and ^{13}C resonances of **8** were assigned as shown in Tables 1 and 2. A detailed NMR comparison of **8**, **7**, and **5** indicated that compound **8** had the same aglycone as **7** and the same saccharide chain as **5**, which was further confirmed by NMR experiments. In the ^{13}C NMR spectra, the most critical differences between compounds **8** and **5** were at C-2 (δ_C 57.9 and 61.7), C-17 (δ_C 70.0 and 62.2), and C-7 (δ_C 75.6 and 73.8), which indicated that **8** and **5** were also diastereomers.⁴ Compound **8** was thus established as α -sulfoxidebreyningenin 3-*O*-[α -rhamnopyranosyl-(1 \rightarrow 3)]- α -rhamnopyranosyl-(1 \rightarrow 2)- β -glucopyranosyl-(1 \rightarrow 2)- β -quinovopyranoside (breyinin G).

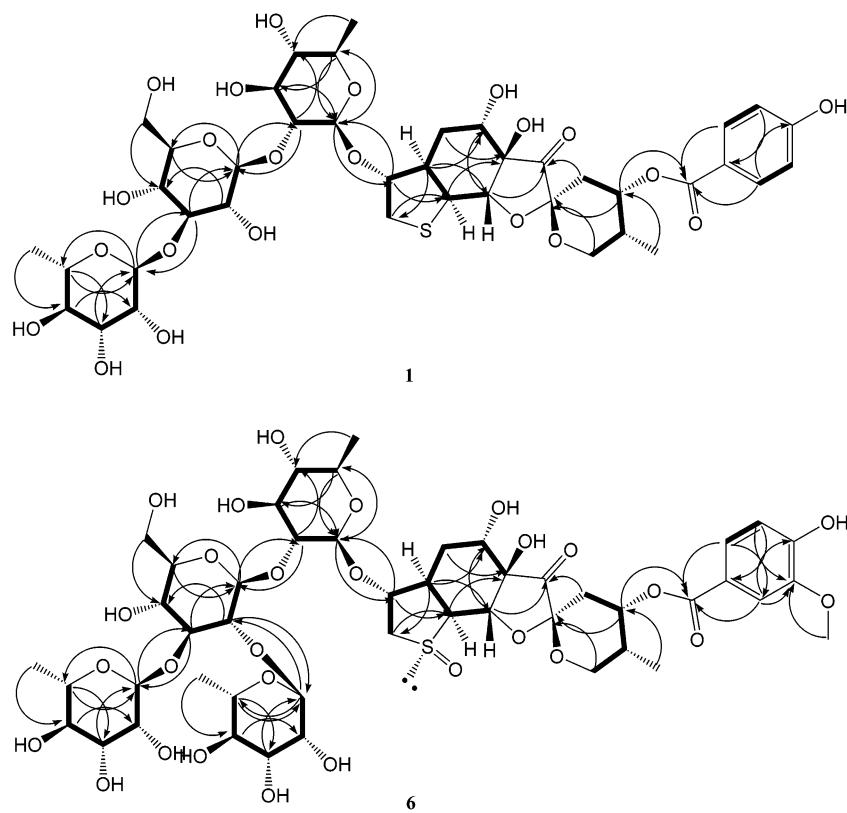


Figure 1. ^1H – ^1H COSY (—) and key HMBC correlations ($^1\text{H} \curvearrowright ^{13}\text{C}$) of compounds **1** and **6**.

Two known sulfur-containing spiroketal glycosides, **9** (epibreynin B) and **10** (breynin B), were also isolated from *B. fruticosa*. Although compound **9** was previously reported as a semisynthetic oxidation product,⁴ this is the first time that it has been isolated from a natural source. Compounds **9** and **10** could be obtained by oxidation of breynin A with Davis' phenyloxaziridine in a 1:7 ratio.⁴ Although compounds **2**–**10** containing sulfoxide functions may be oxidative artifacts during the isolation process, they may also be oxidized by the oxidase in the plant. Natural products with sulfoxide and sulfone groups have also been reported from *Allium* species.^{15–18} The skeleton of the aglycones of **1**–**10** could be classified into the sesquiterpenoid category; however, sulfur-containing sesquiterpenes from terrestrial resources are relatively rare.

Experimental Section

General Experimental Procedures. Optical rotations were measured with a Perkin-Elmer 241MC polarimeter. IR spectra were recorded using a Perkin-Elmer 577 spectrometer. LR-ESIMS were measured using a Finnigan LCQ-DECA instrument, and HR-ESIMS data were obtained on a Mariner spectrometer. NMR spectra were recorded on a Bruker AM 400 or INOVA-600 spectrometer with TMS as internal standard, and chemical shifts are expressed in δ ppm. Preparative HPLC was carried out using a Varian SD-1 instrument, equipped with Merck NW25 C₁₈ column (10 μm , 20 mm \times 250 mm), and Prostar 320 UV/vis detector. Column chromatographic separations were carried out by using poroporous resin D-101 (Huazhen Corporation of Science and Technology, Shanghai, China), silica gel H60 (300–400 mesh) (Qingdao Haiyang Chemical Group Corporation, Qingdao, China), and Sephadex LH-20 (Pharmacia Biotech AB, Uppsala, Sweden) as packing materials. HSGF254 silica gel TLC plates (Yantai Chemical Industrial Institute, Yantai, China) and RP-18 WF₂₅₄ TLC plates (Merck) were used for analytical TLC.

Plant Material. The aerial parts of *B. fruticosa* were collected in the suburb of Nanning, Guangxi Province, People's Republic of China, in June 2005 and identified by Prof. Ding Fang of Guangxi Institute of Traditional Chinese Medicine. A voucher specimen (No. SIMMS0506) is deposited in the Herbarium of Shanghai Institute of Materia Medica, Chinese Academy of Sciences.

Extraction and Isolation. The aerial parts of *B. fruticosa* (30.0 kg) were powdered and percolated at room temperature with 95% EtOH (40 L \times 3). The filtrate was concentrated to dryness *in vacuo*, suspended in 20% EtOH, and filtered. After evaporation of EtOH from the filtrate, the aqueous residue (10 L) was extracted with CHCl₃ and *n*-BuOH (10 L \times 3 each), successively. The *n*-BuOH fraction (800 g) was subjected to a column of Diaion D-101 and eluted with H₂O and 25, 50, 75, and 100% MeOH. The 75% MeOH fraction (40 g) was further separated by VLC over a silica gel column using CHCl₃/MeOH (10:1, 8:1, 6:1, 4:1, 2:1, and 1:1) to give five subfractions (A–E). Fraction D (4.96 g) was chromatographed on a silica gel column eluted with a gradient of CHCl₃/MeOH (3:1 and 1:1) to give four subfractions (D₁–D₄). Fraction D₃ (432 mg) was further subjected to HPLC eluted with MeOH/H₂O (10% to 70% MeOH within 90 min) and then over Sephadex LH-20 CC with 95% EtOH to yield **1** (18 mg). Further purification of subfraction D₄ (170 mg), by HPLC using MeOH/H₂O (10% to 70% MeOH within 90 min) as eluent, afforded four subfractions, D₄₁–D₄₂. Fraction D₄₁ was further purified by PTLC (developed with *n*-BuOH/HOAc/H₂O, 6:1:1) and Sephadex LH-20 CC using 95% EtOH as eluent to afford **10** (10 mg). Fraction E (11.6 g) was subjected to Sephadex LH-20 CC with 95% EtOH to give two subfractions, E₁ and E₂. Fraction E₁ (1.20 g) was chromatographed on a silica gel column eluting with a CHCl₃/MeOH/H₂O (3:1:0.1) gradient to yield **2** (137 mg) and another two subfractions, E₁₁ and E₁₂. Compounds **3** (7.0 mg) and **4** (9.0 mg) were obtained from fraction E₁₁ (67 mg) by PTLC (developed with *n*-BuOH/HOAc/H₂O, 6:1:1) and Sephadex LH-20 CC with 95% EtOH. Fraction E₁₂ (320 mg) was further subjected to HPLC eluted with MeOH/H₂O (10% to 30% MeOH within 20 min and 30% to 70% MeOH within 70 min) to yield **5** (85 mg), **6** (14 mg), **7** (20 mg), **8** (16 mg), and fraction E₁₂₁. From fraction E₁₂₁ (17 mg), **9** (11 mg) was obtained using PTLC (developed with *n*-BuOH/HOAc/H₂O, 6:1:1) and was purified by Sephadex LH-20 CC with 95% EtOH.

Breynin C (1): amorphous powder; $[\alpha]^{24}_D +7.3$ (*c* 0.60, MeOH); IR (KBr) ν_{max} 3420, 2933, 1782, 1691, 1608, 1516, 1281, 1167, 1072, 773 cm⁻¹; ^{13}C and ^1H NMR data, see Tables 1 and 2; ESIMS (positive-ion mode) m/z 943.6 [M + Na]⁺; HRESIMS m/z 921.3094 [M + H]⁺ (calcd for C₄₀H₅₇O₂₂S, 921.3062).

Epibreynin D (2): amorphous powder; $[\alpha]^{24}_D -1.4$ (*c* 0.57, MeOH); IR (KBr) ν_{max} 3415, 2933, 1784, 1697, 1608, 1516, 1279, 1165, 1070,

773 cm⁻¹; ¹³C and ¹H NMR data, see Tables 1 and 2; ESIMS (positive-ion mode) *m/z* 959.2 [M + Na]⁺; HRESIMS *m/z* 959.2845 [M + Na]⁺ (calcd for C₄₀H₅₆O₂₃SNa, 959.2831).

Epibreyinin E (3): colorless gum; [α]_D²⁴ +4.4 (*c* 0.34, MeOH); IR (film) ν_{max} 3410, 2921, 1784, 1693, 1610, 1516, 1279, 1165, 1072, 760 cm⁻¹; ¹³C and ¹H NMR data, see Tables 1 and 2; ESIMS (positive-ion mode) *m/z* 813.1 [M + Na]⁺; HRESIMS *m/z* 813.2257 [M + Na]⁺ (calcd for C₃₄H₄₆O₁₉SNa, 813.2252).

Epibreyinin F (4): colorless gum; [α]_D²⁴ -1.3 (*c* 0.48, MeOH); IR (film) ν_{max} 3410, 2920, 1784, 1692, 1610, 1516, 1279, 1165, 1072, 760 cm⁻¹; ¹³C and ¹H NMR data, see Tables 1 and 2; ESIMS (positive-ion mode) *m/z* 945.1 [M + Na]⁺; HRESIMS *m/z* 945.2706 [M + Na]⁺ (calcd for C₃₉H₅₄O₂₃SNa, 945.2674).

Epibreyinin G (5): amorphous powder; [α]_D²⁴ -7.7 (*c* 0.76, MeOH); IR (KBr) ν_{max} 3406, 2933, 1784, 1694, 1608, 1514, 1279, 1167, 1070, 774 cm⁻¹; ¹³C and ¹H NMR data, see Tables 1 and 2; ESIMS (positive-ion mode) *m/z* 1105.2 [M + Na]⁺; HRESIMS *m/z* 1105.3418 [M + Na]⁺ (calcd for C₄₆H₆₆O₂₇SNa, 1105.3410).

Epibreyinin H (6): colorless gum; [α]_D²⁴ -9.6 (*c* 0.80, MeOH); IR (KBr) ν_{max} 3419, 2933, 1784, 1697, 1599, 1512, 1284, 1168, 1074, 1043, 770 cm⁻¹; ¹³C and ¹H NMR data, see Tables 1 and 2; ESIMS (positive-ion mode) *m/z* 1135.2 [M + Na]⁺; HRESIMS *m/z* 1135.3535 [M + Na]⁺ (calcd for C₄₇H₆₈O₂₈SNa, 1135.3516).

Breynin D (7): colorless gum; [α]_D²⁴ -10.0 (*c* 0.62, MeOH); IR (film) ν_{max} 3385, 2933, 1782, 1691, 1610, 1516, 1277, 1167, 1072, 773 cm⁻¹; ¹³C and ¹H NMR data, see Tables 1 and 2; ESIMS (positive-ion mode) *m/z* 959.2 [M + Na]⁺; HRESIMS *m/z* 959.2850 [M + Na]⁺ (calcd for C₄₀H₅₆O₂₃SNa, 959.2831).

Breynin G (8): colorless gum; [α]_D²⁴ -17.5 (*c* 0.87, MeOH); IR (KBr) ν_{max} 3415, 2933, 1782, 1691, 1608, 1514, 1279, 1167, 1074, 772 cm⁻¹; ¹³C and ¹H NMR data, see Tables 1 and 2; ESIMS (positive-ion mode) *m/z* 1105.2 [M + Na]⁺; HRESIMS *m/z* 1105.3359 [M + Na]⁺ (calcd for C₄₆H₆₆O₂₇SNa, 1105.3410).

Acetylation of Epibreyinin D (2). Ac₂O (1.0 mL) was added to a solution of epibreyinin D (2) (4.0 mg) in pyridine (1.0 mL). The mixture was allowed to stir at room temperature for 48 h. Following concentration *in vacuo* and purification by PTLC (developed with CHCl₃), the undecaacetate **2a** was obtained (2.3 mg).

Compound 2a: amorphous powder; [α]_D²⁴ -4.0 (*c* 0.10, MeOH); ESIMS (positive-ion mode) *m/z* 1379.2 [M + Na]⁺; ¹H NMR (600 MHz, CDCl₃) 8.10 (2H, d, *J* = 8.8 Hz, H-21/25), 7.15 (2H, d, *J* = 8.8 Hz, H-22/24), 5.44 (1H, m), 5.04 (1H, t, *J* = 9.3 Hz, H-3_{qui}), 4.70–5.04 (6H, m, H-6, H-2_{glc}, H-4_{glc}, H-2_{rha}, H-3_{rha}, H-4_{rha}), 4.72 (1H, br s, H-1_{rha}), 4.61 (1H, br d, *J* = 4.4 Hz, H-16), 4.57 (1H, t, *J* = 9.8 Hz, H-4_{qui}), 4.49 (1H, d, *J* = 7.8 Hz, H-1_{glc}), 4.36 (1H, d, *J* = 7.8 Hz, H-1_{qui}), 3.98–4.48 (4H, m, H-2, H-6_{glc}), 3.93 (1H, dd, *J* = 11.3, 7.3 Hz, H-13_α), 3.89 (1H, br d, *J* = 5.8 Hz, H-3), 3.78–3.85 (2H, m, H-13_β, H-5_{rha}), 3.65 (1H, t, *J* = 9.2 Hz, H-3_{glc}), 3.58 (1H, t, *J* = 9.3 Hz, H-2_{qui}), 3.40–3.46 (2H, m, H-5_{qui}, H-5_{glc}), 3.10 (1H, dd, *J* = 9.3, 5.4 Hz, H-17), 2.80 (1H, m, H-4), 2.34 (3H, s, CH₃CO), 2.19–2.30

(3H, m, H-10, H-12), 2.15 (3H, s, CH₃CO), 2.10 (3H, s, CH₃CO), 2.09 (3H, s, CH₃CO), 2.08 (3H, s, CH₃CO), 2.05 (3H, s, CH₃CO), 2.01 (3H, s, CH₃CO), 2.00 (3H, s, CH₃CO), 1.97 (3H, s, CH₃CO), 1.95 (1H, m, H-5_β), 1.91 (3H, s, CH₃CO), 1.85 (1H, m, H-5_α), 1.09 (3H, d, *J* = 6.4 Hz, H-6_{rha}), 1.06 (3H, d, *J* = 6.3 Hz, H-6_{qui}), 1.02 (3H, d, *J* = 6.8 Hz, H-18).

Acknowledgment. The authors are grateful to Shanghai Committee of Science and Technology for partial financial support (No. 03DZ19228) and wish to thank Prof. Y. Zhou of Guangxi Institute of Traditional Chinese Medicine for offering the plant material.

Supporting Information Available: 1D and 2D NMR spectra of compound **5**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References and Notes

- Koshiyama, H.; Hatori, M.; Ohkuma, H.; Sakai, F.; Imanishi, H.; Ohbayashi, M.; Kawaguchi, H. *Chem. Pharm. Bull.* **1976**, 24, 169–172.
- Ohkuma, H.; Tsuno, T.; Konishi, M.; Naito, T.; Kawaguchi, H. *Chem. Pharm. Bull.* **1991**, 39, 942–944.
- Smith, A. B., III; Gallagher, R. T.; Keenan, T. P.; Furst, G. T.; Dormer, P. G. *Tetrahedron Lett.* **1991**, 32, 6847–6850.
- Smith, A. B., III; Keenan, T. P.; Gallagher, R. T.; Furst, G. T.; Dormer, P. G. *J. Org. Chem.* **1992**, 57, 5115–5120.
- Lajis, N. H.; Retnam, A.; Hassan, H. A.; Khan, M. N.; Chua, C. H.; Sargent, M. V. *Indian J. Chem.* **1997**, 36B, 206–207.
- Lajis, N. H.; Guan, O. B.; Sargent, M. V.; Skelton, B. W.; White, A. H. *Aust. J. Chem.* **1992**, 45, 1893–1897.
- Morikawa, H.; Kasai, R.; Otsuka, H.; Hirata, E.; Shinzato, T.; Aramoto, M.; Takeda, Y. *Chem. Pharm. Bull.* **2004**, 52, 1086–1090.
- Fu, G. M.; Xu, Z. L.; Yu, B. Y.; Zhu, D. N. *China J. Chin. Mater. Med.* **2004**, 29, 1052–1054.
- Fu, G. M.; Xu, Z. L.; Yu, B. Y.; Zhu, D. N. *J. China Pharm. Univ.* **2004**, 35, 114–116.
- Sasaki, K.; Hirata, Y. *Tetrahedron Lett.* **1973**, 14, 2439–2442.
- Sakai, F.; Ohkuma, H.; Koshiyama, H.; Naito, T.; Kawaguchi, H. *Chem. Pharm. Bull.* **1976**, 24, 119–120.
- Agrawal, P. K. *Phytochemistry* **1992**, 31, 3307–3330.
- Smith, A. B., III; Empfield, J. R.; Rivero, R. A.; Vaccaro, H. A.; Duan, J. J.-W.; Sulikowski, M. M. *J. Am. Chem. Soc.* **1992**, 114, 9419–9434.
- Su, Y. F.; Koike, K.; Nikaido, T.; Liu, J. S.; Zheng, J. H.; Guo, D. *J. Nat. Prod.* **2003**, 66, 1593–1599.
- Stoll, A.; Seebach, E. *Helv. Chim. Acta* **1949**, 32, 197–205.
- Yang, Z. N.; Yang, X. S. *J. Guizhou Norm. Univ.* **2004**, 22, 104–112.
- Hu, Q. H.; Yang, Q.; Yamato, O.; Yamasaki, M.; Maede, Y.; Yoshihara, T. *J. Agric. Food. Chem.* **2002**, 50, 1059–1062.
- Bayer, T. H.; Breu, W.; Seligmann, O.; Wray, V.; Wagner, H. *Phytochemistry* **1989**, 28, 2373–2378.

NP0606300